## CHART - 4 > Important Chemicals & Compounds

ACIDS		BASES	
<ol> <li>MURIATIC ACID</li> <li>AQUA FORTIS</li> <li>OIL OF VITRIOL</li> <li>AQUA REGIA</li> <li>VINEGAR</li> </ol>	Hydrochloric acid [HCl] Nitric acid [HNO <sub>3</sub> ] Sulphuric acid [H <sub>2</sub> SO <sub>4</sub> ] Nitric & hydrochloric acid [conc. – 1 vol. & 3 vol.] Acetic acid [CH <sub>3</sub> COOH]	<ol> <li>CAUSTIC POTASH</li> <li>CAUSTIC SODA</li> <li>LIQUOR AMMONIA</li> </ol>	Potassium hydroxide Sodium hydroxide Ammonium hydroxide
SALTS		ORES	
10. WASHING SODA  11. BAKING SODA  12. GLAUBER'S SALT  13. LUNAR CAUSTIC  14. CINNABAR  15. BLUE VITRIOL  16. GREEN VITRIOL	Acid salt [NaHSO <sub>4</sub> ]  Mixed salt [CaOCl <sub>2</sub> ]  Complex salt [Na [Ag (CN) <sub>2</sub> ]  Sodium chloride [NaCl]  Ammonium chloride  Silver chloride [AgCl]  Potassium nitrate [KNO <sub>3</sub> ]  Sodium carbonate [NaNO <sub>3</sub> ]  Sodium carbonate [hyd.]- [Na <sub>2</sub> CO <sub>3</sub> .10H <sub>2</sub> O ]  Sodium bicarbonate- [NaHCO <sub>3</sub> ]  Sodium sulphate [hyd.]- [Na <sub>2</sub> SO <sub>4</sub> .10H <sub>2</sub> O]  Silver nitrate [AgNO <sub>3</sub> ]  Mercuric sulphide [HgS]  Copper sulphate [hyd.]- [CuSO <sub>4</sub> .5H <sub>2</sub> O]  Iron [II] sulphate [hyd.]- [FeSO <sub>4</sub> .7H <sub>2</sub> O]	ORES OF ALUMINIUM  1. BAUXITE  2. CRYOLITE  ORES OF ZINC  3. ZINC BLENDE  4. CALAMINE  5. ZINCITE  ORES OF IRON  6. HAEMATITE  7. MAGNETITE  8. IRON PYRITES  ORE OF LEAD  9. GALENA  ORE OF COPPER  10. CUPRITE	Aluminium oxide [hyd.]- [Al <sub>2</sub> O <sub>3</sub> .2H <sub>2</sub> O] Sodium aluminium fluoride - [Na <sub>3</sub> AlF <sub>6</sub> ]  Zinc sulphide - [ZnS] Zinc carbonate - [ZnCO <sub>3</sub> ] Zinc oxide - [ZnO]  Iron [III] oxide- [Fe <sub>2</sub> O <sub>3</sub> ] Tri iron tetraoxide - [Fe <sub>3</sub> O <sub>4</sub> ] Iron sulphide [FeS <sub>2</sub> ]  Lead sulphide - [PbS]
17. WHITE VITRIOL	Zinc sulphate [hyd] - [ZnSO <sub>4</sub> .7H <sub>2</sub> O]	HYDROCARBONS -	
18. EPSOM SALT  19. ALUM  20. LITHARGE  21. SILICA  22. PYROLUSITE	Magnesium sulphate[hyd] [MgSO <sub>4</sub> .7H <sub>2</sub> O]  Potassium aluminium sulphate [K <sub>2</sub> SO <sub>4</sub> .Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> .24 H <sub>2</sub> O]  Lead monoxide [PbO]  Silicon dioxide [SiO <sub>2</sub> ]  Manganese [IV] oxide -	<ol> <li>ETHANE</li> <li>ETHENE [ethylene]</li> <li>ETHYNE [acetylene]</li> <li>METHANOL [methyl alcohol]</li> </ol>	CH <sub>3</sub> OH
22. PYROLUSITE	Manganese [IV] oxide - [MnO <sub>2</sub> ]	6. ETHANOL [ethyl alcohol]	C <sub>2</sub> H <sub>5</sub> OH